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ANSWER Write the answers to the  
following questions in the space  
provided. 22.

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Chemistry (4th Edition) Burdge, Julia

Publisher McGraw-Hill Publishing

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CHAPTER 9 REVIEW Stoichiometry

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SECTION 3 PROBLEMS Write the answer on the line to the left. Show all your work in the space provided.

1. 88% The actual yield of a reaction is 22 g and the theoretical yield is 25 g. Calculate the percentage yield.

2. 6.0 mol of  $N_2$  are mixed with 12.0 mol of H

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the following questions in the space provided. 1. b A balanced chemical equation represents all the following except (a) experimentally established facts. (b) the mechanism by which reactants combine to form ...

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CHAPTER 3 REVIEW Atoms: The Building  
Blocks of Matter SECTION 3 SHORT  
ANSWER Answer the following questions  
in the space provided. 1. Explain the  
difference between the mass  
number and the atomic number of a

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nuclide. Mass number is the total number of protons and neutrons in the nucleus of an isotope.

## **3 Atoms: The Building Blocks of Matter**

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